

SHORT  
COMMUNICATIONS

## New P-Chiral Polyfluoroalkyl Phosphorodiamidite Ligand in Asymmetric Transformations Catalyzed by Palladium and Copper Complexes

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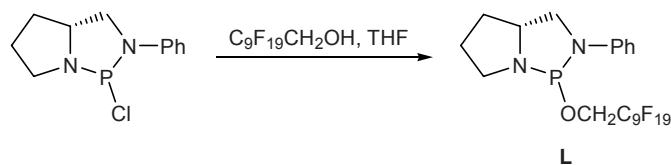
Chiral phosphorus-containing ligands having polyfluoroalkyl substituents may be used for the preparation of metal complex catalysts which could be recycled via phase separation [1]. Following the approach proposed by us previously [2], we synthesized a new P-chiral polyfluoroalkyl phosphorodiamidite **L** as shown in Scheme 1. The product was stable on storage and readily soluble in organic solvents. It was tested as chiral auxiliary in palladium-catalyzed enantioselective amination of 1,3-diphenylprop-2-en-1-yl acetate (**I**) with diisopropylamine according to the procedure described in [3] (Scheme 2). In all experiments, a steadily high enantioselectivity level was reached (*ee* 91–95%), regardless of the **L**/Pd molar ratio (1:1 or 2:1) and reaction medium. However, the substrate conversion turned out to be quite sensitive to the solvent

nature: it did not exceed 32% in tetrahydrofuran but was almost complete in methylene chloride.

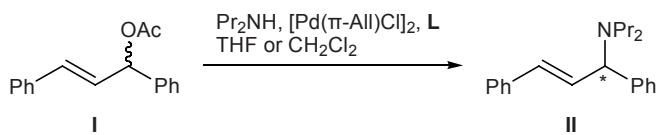
Ligand **L** was also used in copper-catalyzed conjugate addition of diethylzinc to cyclohex-2-en-1-one (**III**) [4] (Scheme 3). The reaction was carried out in diethyl ether in the presence of copper thiophene-carboxylate as pre-catalyst, and the molar ratio **L**–Cu was 2:1 (−30°C, 3 h). The optical yield of 3-ethylcyclohexanone (**IV**) was 70% (*ee*), the substrate conversion being higher than 99%.

Compound **L** is a fairly effective stereoinductor. It is inferior to phosphoramidites based on biphenyl-2,2'-diol and BINOL (1,1'-binaphthalene-2,2'-diol) in Cu-catalyzed conjugate addition reactions [4], but it ensures higher enantioselectivity in Pd-catalyzed allyla-

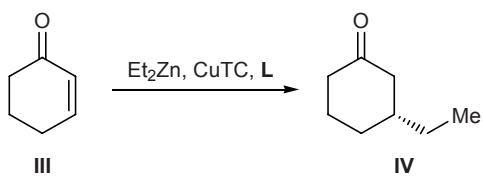
Scheme 1.



Scheme 2.



Scheme 3.



tion of 1,3-diphenylprop-2-en-1-yl acetate [5, 6]. On the other hand, ligand **L** was much more effective than its analogs having a 3-phenyl-1,3-diaza-2-phosphabicyclo[3.3.0]octane skeleton in Cu-catalyzed conjugate addition of  $\text{Et}_2\text{Zn}$  to cyclohex-2-en-1-one. The corresponding  $\pi^*$ ,N-bidentate phosphorodiamidites ensured enantiomeric excess of no higher than 55% [7], while  $\pi^*$ -monodentate, only *ee* 10–20% [8].

**(2S,5R)-2-(2,2,3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,10-Nonadecafluorodecyloxy)-3-phenyl-1,3-diaza-2-phosphabicyclo[3.3.0]octane (L).** 2,2,3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,10-Nonadecafluorodecan-1-ol, 0.785 g (1.57 mmol), was added under vigorous stirring at 20°C to a solution of 0.378 g (1.57 mmol) of (2S,5R)-2-chloro-3-phenyl-1,3-diaza-2-phosphabicyclo[3.3.0]-octane and 0.22 ml (1.57 mmol) of triethylamine in 12 ml of THF. The resulting solution was stirred for 30 min at 20°C, heated to the boiling point, and kept boiling for 2 h. It was then cooled to 20°C, the precipitate of triethylamine hydrochloride was filtered off, the filtrate was evaporated under reduced pressure (40 mm), and the residue was extracted with hexane (3×15 ml). The extracts were combined, filtered, and evaporated under reduced pressure (40 mm), and the residue was evacuated for 2 h at a residual pressure of 1 mm. Yield 0.896 g (81%), light yellow oily substance which solidified on storage, mp 62–63°C.  $^{13}\text{C}$  NMR spectrum ( $\text{CDCl}_3$ ),  $\delta_{\text{C}}$ , ppm: 145.1 d ( $\text{C}_{\text{arom}}$ ,  $^2J = 16.8$  Hz), 129.2 s ( $\text{CH}_{\text{arom}}$ ), 119.7 s ( $\text{CH}_{\text{arom}}$ ), 115.0 d ( $\text{CH}_{\text{arom}}$ ,  $^3J = 11.7$  Hz), 109.4 br.m ( $\text{C}^2-\text{C}^{10'}$ ), 63.2 d ( $\text{C}^5$ ,  $^2J = 8.8$  Hz), 58.6 m ( $\text{OCH}_2$ ,  $^2J = 5.3$  Hz), 54.9 d ( $\text{C}^4$ ,  $^2J = 6.6$  Hz), 48.3 d ( $\text{C}^8$ ,  $^2J = 37.2$  Hz), 31.8 s ( $\text{C}^6$ ), 26.4 d ( $\text{C}^7$ ,  $^3J = 3.7$  Hz).  $^{31}\text{P}$  NMR spectrum ( $\text{CDCl}_3$ ):  $\delta_{\text{P}}$  124.1 ppm.  $^{19}\text{F}$  NMR spectrum ( $\text{CDCl}_3$ ),  $\delta_{\text{F}}$ , ppm: -79.1 to -79.35 m (3F), -117.92 s (2F), -120.25 s (8F), -121.15 s (2F), -121.4 s (2F), -124.55 s (2F). Mass spectrum (electron impact, 70 eV),  $m/z$  ( $I_{\text{rel}}$ , %): 704 [ $M]^+$  (3), 484 [ $\text{C}_{10}\text{H}_{2}\text{F}_{19} + \text{H}]^+$  (8), 222 [ $M - \text{C}_{10}\text{H}_{2}\text{F}_{19} + \text{H}]^+$  (100), 205 [ $M - \text{C}_{10}\text{H}_{2}\text{F}_{19}\text{O}]^+$  (82). Found, %: C 36.04; H 2.37; N 3.79.  $\text{C}_{21}\text{H}_{16}\text{F}_{19}\text{N}_2\text{OP}$ . Calculated, %: C 35.81; H 2.29; N 3.98.

**Palladium-catalyzed allylic amination of 1,3-diphenylprop-2-en-1-yl acetate (I) with dipropylamine.** A solution of 3.7 mg (0.01 mmol) of  $[\text{Pd}(\pi\text{-All})\text{Cl}]_2$  and 7.04–14.08 mg (0.01–0.02 mmol) of ligand **L** in 5 ml of THF or methylene chloride was stirred for 10 min. Compound **I**, 0.1 ml (0.5 mmol), was then added, the mixture was stirred for 15 min, 0.2 ml (1.5 mmol) of freshly distilled dipropylamine was added, and the mixture was stirred for 48 h and passed through a layer of silica gel. The solvent was removed under reduced pressure (40 mm), and the residue was kept for 1 h at a residual pressure of 1 mm. The *ee* value was determined by HPLC (Daicel Chiralcel OD-H;  $\text{C}_6\text{H}_{14}-i\text{-PrOH}-\text{HNEt}_2$ , 1000:1:1, 0.5 ml/min;  $\lambda$  254 nm) as described in [9].

**Copper-catalyzed conjugate addition of diethylzinc to 2-cyclohexen-1-one (III).** Ligand **L**, 0.04 mmol, was added at 20°C under argon to a solution of 0.02 mmol of copper thiophenecarboxylate in 2.5 ml of diethyl ether. The mixture was stirred for 30 min, cooled to -30°C, and kept for 30 min at that temperature. A 1 M solution of diethylzinc in hexane, 2.4 mmol, was added, the mixture was stirred for 20 min, and a solution of 2 mmol of compound **III** in 0.5 ml of diethyl ether was added. The mixture was stirred for 3 h, allowed to warm up to room temperature, diluted with 20 ml of diethyl ether, and extracted with 2 N hydrochloric acid. The organic phase was then washed with a concentrated solution of sodium chloride, dried over sodium sulfate, and evaporated under reduced pressure (40 mm). The residue was purified by flash chromatography on silica gel using cyclohexane–ethyl acetate (1:1) as eluent. The *ee* value was determined by HPLC as described in [4].

All reactions were carried out under dry argon in thoroughly dehydrated solvents. The  $^{31}\text{P}$ ,  $^{13}\text{C}$ , and  $^{19}\text{F}$  NMR spectra were recorded on a Bruker Avance-400 spectrometer at 161.98, 100.61, and 282.4 MHz, respectively, using 85%  $\text{H}_3\text{PO}_4$  in  $\text{D}_2\text{O}$ ,  $\text{CDCl}_3$  ( $\delta_{\text{C}}$  76.91 ppm), and  $\text{CCl}_3\text{F}$  as references. Signals in the  $^{13}\text{C}$  NMR spectrum of ligand **L** were assigned using DEPT pulse sequence. The mass spectrum was obtained on a Varian MAT-311 spectrometer. Elemental analysis was performed at the Organic Microanalysis Laboratory, Nesmeyanov Institute of Organometallic Compounds, Russian Academy of Sciences.

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